



calculatoratoz.com



unitsconverters.com

Refrigeration and Air Conditioning Formulas

Calculators!

Examples!

Conversions!

Bookmark calculatoratoz.com, unitsconverters.com

Widest Coverage of Calculators and Growing - **30,000+ Calculators!**
Calculate With a Different Unit for Each Variable - **In built Unit Conversion!**
Widest Collection of Measurements and Units - **250+ Measurements!**

Feel free to SHARE this document with your friends!

[Please leave your feedback here...](#)



List of 25 Refrigeration and Air Conditioning Formulas

Refrigeration and Air Conditioning

Air Refrigeration Cycles

1) Energy Performance Ratio of Heat Pump

$$\text{fx } \text{COP}_{\text{theoretical}} = \frac{Q_{\text{delivered}}}{W_{\text{per min}}}$$

[Open Calculator !\[\]\(de95854c7ee024cfadc48187bbb781b2_img.jpg\)](#)

$$\text{ex } 4.807692 = \frac{1250\text{kJ}/\text{min}}{260\text{kJ}/\text{min}}$$

2) Relative Coefficient of Performance

$$\text{fx } \text{COP}_{\text{relative}} = \frac{\text{COP}_{\text{actual}}}{\text{COP}_{\text{theoretical}}}$$

[Open Calculator !\[\]\(6a9b39b98eb945faa14c645ec99e4eaa_img.jpg\)](#)

$$\text{ex } 0.833333 = \frac{5}{6}$$



3) Theoretical Coefficient of Performance of Refrigerator

$$\text{fx } \text{COP}_{\text{theoretical}} = \frac{Q}{W}$$

[Open Calculator !\[\]\(cbe80b694ebd74fcfe136a095b608235_img.jpg\)](#)

$$\text{ex } 1.5 = \frac{600\text{kJ/kg}}{400\text{kJ/kg}}$$

Bell-Coleman Cycle or Reversed Brayton or Joule Cycle

4) Compression or Expansion Ratio

$$\text{fx } r_p = \frac{P_2}{P_1}$$

[Open Calculator !\[\]\(5361750c22c4e047a52f4eac1ec2d4cc_img.jpg\)](#)

$$\text{ex } 2.5 = \frac{10\text{Bar}}{4\text{Bar}}$$

5) COP of Bell-Coleman Cycle for given Compression Ratio and Adiabatic Index

$$\text{fx } \text{COP}_{\text{theoretical}} = \frac{1}{r_p^{\frac{\gamma-1}{\gamma}} - 1}$$

[Open Calculator !\[\]\(b792654f2cef9719eabeb6c5be00811e_img.jpg\)](#)

$$\text{ex } 4.565925 = \frac{1}{(2)^{\frac{1.4-1}{1.4}} - 1}$$



6) COP of Bell-Coleman Cycle for given Temperatures, Polytropic Index and Adiabatic Index

fx

Open Calculator 

$$\text{COP}_{\text{theoretical}} = \frac{T_1 - T_4}{\left(\frac{n}{n-1}\right) \cdot \left(\frac{\gamma-1}{\gamma}\right) \cdot ((T_2 - T_3) - (T_1 - T_4))}$$

ex

$$0.538462 = \frac{300\text{K} - 290\text{K}}{\left(\frac{1.30}{1.30-1}\right) \cdot \left(\frac{1.4-1}{1.4}\right) \cdot ((350\text{K} - 325\text{K}) - (300\text{K} - 290\text{K}))}$$

7) Heat Absorbed during Constant Pressure Expansion Process

fx

$$Q_{\text{Absorbed}} = C_p \cdot (T_1 - T_4)$$

Open Calculator 

ex

$$10.05\text{kJ/kg} = 1.005\text{kJ/kg}\cdot\text{K} \cdot (300\text{K} - 290\text{K})$$

8) Heat Rejected during Constant pressure Cooling Process

fx

$$Q_R = C_p \cdot (T_2 - T_3)$$

Open Calculator 

ex

$$25.125\text{kJ/kg} = 1.005\text{kJ/kg}\cdot\text{K} \cdot (350\text{K} - 325\text{K})$$



Air Refrigeration Systems

9) Initial Mass of Evaporant Required to be Carried for given Flight Time

$$\text{fx } M = \frac{Q_r \cdot t}{h_{fg}}$$

[Open Calculator !\[\]\(74d4806277d7e73349d8e8c0897931e9_img.jpg\)](#)

$$\text{ex } 0.442478\text{kg} = \frac{50\text{kJ}/\text{min} \cdot 20\text{min}}{2260\text{kJ}/\text{kg}}$$

10) Local Sonic or Acoustic Velocity at Ambient Air Conditions

$$\text{fx } a = \left(\gamma \cdot [R] \cdot \frac{T_i}{MW} \right)^{0.5}$$

[Open Calculator !\[\]\(8bba887393ca45b761e5cb49e755e762_img.jpg\)](#)

$$\text{ex } 172.0047\text{m/s} = \left(1.4 \cdot [R] \cdot \frac{305\text{K}}{0.120\text{kg}} \right)^{0.5}$$

11) Ram Efficiency

$$\text{fx } \eta = \frac{(P_2') - P_i}{P_f - P_i}$$

[Open Calculator !\[\]\(0fb13ad0bfa3d86868cdd3883e5665b3_img.jpg\)](#)

$$\text{ex } 0.866667 = \frac{150000\text{Pa} - 85000\text{Pa}}{160000\text{Pa} - 85000\text{Pa}}$$



Simple Air Cooling System

12) Specific Heat Capacity at Constant Pressure using Adiabatic Index

$$\text{fx } C_p = \frac{\gamma \cdot [R]}{\gamma - 1}$$

[Open Calculator !\[\]\(950a62bbddad88d64435fd35607dfc42_img.jpg\)](#)

$$\text{ex } 0.029101 \text{kJ/kg} \cdot \text{K} = \frac{1.4 \cdot [R]}{1.4 - 1}$$

13) Temperature Ratio at Start and End of Ramming Process

$$\text{fx } T_{\text{ratio}} = 1 + \frac{v_{\text{process}}^2 \cdot (\gamma - 1)}{2 \cdot \gamma \cdot [R] \cdot T_i}$$

[Open Calculator !\[\]\(73002692dd5e7a64e60946be3158e719_img.jpg\)](#)

$$\text{ex } 1.202801 = 1 + \frac{(60 \text{m/s})^2 \cdot (1.4 - 1)}{2 \cdot 1.4 \cdot [R] \cdot 305 \text{K}}$$

Basics of Refrigeration and Air Conditioning


14) Entropy Change for Isochoric Process given Pressures

$$\text{fx } \Delta S_{CV} = m_{\text{gas}} \cdot C_{v \text{ molar}} \cdot \ln \left(\frac{P_f}{P_i} \right)$$

[Open Calculator !\[\]\(aab88c0d099e5d18d6533a97b13ec28d_img.jpg\)](#)

$$\text{ex } 130.2996 \text{J/kg} \cdot \text{K} = 2 \text{kg} \cdot 103 \text{J/K} \cdot \text{mol} \cdot \ln \left(\frac{160000 \text{Pa}}{85000 \text{Pa}} \right)$$



15) Entropy Change for Isochoric Process given Temperature 

$$fx \quad \Delta S_{CV} = m_{\text{gas}} \cdot C_{v \text{ molar}} \cdot \ln\left(\frac{T_f}{T_i}\right)$$

Open Calculator 

$$ex \quad 25.38592\text{J/kg}\cdot\text{K} = 2\text{kg} \cdot 103\text{J/K}\cdot\text{mol} \cdot \ln\left(\frac{345\text{K}}{305\text{K}}\right)$$

16) Entropy Change for Isothermal Process given Volumes 

$$fx \quad \Delta S = m_{\text{gas}} \cdot [R] \cdot \ln\left(\frac{V_f}{V_i}\right)$$

Open Calculator 

$$ex \quad 2.77793\text{J/kg}\cdot\text{K} = 2\text{kg} \cdot [R] \cdot \ln\left(\frac{13\text{m}^3}{11\text{m}^3}\right)$$

17) Entropy Change in Isobaric Process given Temperature 

$$fx \quad \Delta S_{CP} = m_{\text{gas}} \cdot C_{p \text{ molar}} \cdot \ln\left(\frac{T_f}{T_i}\right)$$

Open Calculator 

$$ex \quad 30.06876\text{J/kg}\cdot\text{K} = 2\text{kg} \cdot 122\text{J/K}\cdot\text{mol} \cdot \ln\left(\frac{345\text{K}}{305\text{K}}\right)$$

18) Entropy Change in Isobaric Process in Terms of Volume 

$$fx \quad \Delta S_{CP} = m_{\text{gas}} \cdot C_{p \text{ molar}} \cdot \ln\left(\frac{V_f}{V_i}\right)$$

Open Calculator 

$$ex \quad 40.7612\text{J/kg}\cdot\text{K} = 2\text{kg} \cdot 122\text{J/K}\cdot\text{mol} \cdot \ln\left(\frac{13\text{m}^3}{11\text{m}^3}\right)$$



19) Equipment Total Cooling Load 

$$fx \quad Q_T = Q_{\text{per hour}} \cdot L_F$$

[Open Calculator !\[\]\(6605b201d6f14d9b3bcb8ab5f274d107_img.jpg\)](#)

$$ex \quad 10\text{Btu/h} = 8\text{Btu/h} \cdot 1.25$$

20) Heat Transfer at Constant Pressure 

$$fx \quad Q_{\text{per unit}} = m_{\text{gas}} \cdot C_{p \text{ molar}} \cdot (T_f - T_i)$$

[Open Calculator !\[\]\(e8fb589d58dad1692debababa5e928b6_img.jpg\)](#)

$$ex \quad 9.76\text{kJ/kg} = 2\text{kg} \cdot 122\text{J/K} \cdot \text{mol} \cdot (345\text{K} - 305\text{K})$$

21) Isobaric Work for given Mass and Temperatures 

$$fx \quad W_b = N \cdot [R] \cdot (T_f - T_i)$$

[Open Calculator !\[\]\(4688aadfd656ded00cd6bdfae55089a9_img.jpg\)](#)


$$ex \quad 16628.93\text{J} = 50\text{mol} \cdot [R] \cdot (345\text{K} - 305\text{K})$$

22) Isobaric Work for given Pressure and Volumes 

$$fx \quad W_b = P_{\text{abs}} \cdot (V_f - V_i)$$

[Open Calculator !\[\]\(4146d17f71dced09c6ad789cacceaa6d_img.jpg\)](#)

$$ex \quad 200000\text{J} = 100000\text{Pa} \cdot (13\text{m}^3 - 11\text{m}^3)$$

23) Mass Flow Rate in Steady Flow 

$$fx \quad m = A \cdot \frac{u_{\text{Fluid}}}{v}$$

[Open Calculator !\[\]\(08ff79f060f3543d9ed549cc693d8b98_img.jpg\)](#)

$$ex \quad 19.63636\text{kg/s} = 24\text{m}^2 \cdot \frac{9\text{m/s}}{11\text{m}^3/\text{kg}}$$



24) Specific Heat Capacity at Constant Pressure

$$\text{fx } C_{p \text{ molar}} = [R] + C_{v \text{ molar}}$$

[Open Calculator !\[\]\(c3d993ca47bfe2a953c700506ce31fa0_img.jpg\)](#)

$$\text{ex } 111.3145\text{J/K}^*\text{mol} = [R] + 103\text{J/K}^*\text{mol}$$

25) Work Done in Adiabatic Process given Adiabatic Index

$$\text{fx } W = \frac{m_{\text{gas}} \cdot [R] \cdot (T_i - T_f)}{\gamma - 1}$$

[Open Calculator !\[\]\(17413706fd4997a1a4bdf85c6864eee1_img.jpg\)](#)

$$\text{ex } -1662.892524\text{J} = \frac{2\text{kg} \cdot [R] \cdot (305\text{K} - 345\text{K})}{1.4 - 1}$$



Variables Used

- **a** Sonic Velocity (*Meter per Second*)
- **A** Cross Sectional Area (*Square Meter*)
- **C_{p molar}** Molar Specific Heat Capacity at Constant Pressure (*Joule Per Kelvin Per Mole*)
- **C_p** Specific Heat Capacity at Constant Pressure (*Kilojoule per Kilogram per K*)
- **C_{v molar}** Molar Specific Heat Capacity at Constant Volume (*Joule Per Kelvin Per Mole*)
- **COP_{actual}** Actual Coefficient of Performance
- **COP_{relative}** Relative Coefficient of Performance
- **COP_{theoretical}** Theoretical Coefficient of Performance
- **h_{fg}** Latent Heat of Vaporization (*Kilojoule per Kilogram*)
- **L_F** Latent Factor
- **m** Mass Flow Rate (*Kilogram per Second*)
- **M** Mass (*Kilogram*)
- **m_{gas}** Mass of Gas (*Kilogram*)
- **MW** Molecular Weight (*Kilogram*)
- **n** Polytropic Index
- **N** Amount of Gaseous Substance in Moles (*Mole*)
- **P₁** Pressure at Start of Isentropic Compression (*Bar*)
- **p₂'** Stagnation Pressure of System (*Pascal*)
- **P₂** Pressure at End of Isentropic Compression (*Bar*)













- **P_{abs}** Absolute Pressure (*Pascal*)
- **P_f** Final Pressure of System (*Pascal*)
- **P_i** Initial Pressure of System (*Pascal*)
- **Q** Heat Extracted from Refrigerator (*Kilojoule per Kilogram*)
- **$Q_{Absorbed}$** Heat Absorbed (*Kilojoule per Kilogram*)
- **$Q_{delivered}$** Heat Delivered to Hot Body (*Kilojoule per Minute*)
- **$Q_{per\ hour}$** Sensible Cooling Load (*Btu (th) per Hour*)
- **$Q_{per\ unit}$** Heat Transfer (*Kilojoule per Kilogram*)
- **Q_r** Rate of Heat Removal (*Kilojoule per Minute*)
- **Q_R** Heat Rejected (*Kilojoule per Kilogram*)
- **Q_T** Total Cooling Load (*Btu (th) per Hour*)
- **r_p** Compression or Expansion Ratio
- **t** Time in Minutes (*Minute*)
- **T_1** Temperature at Start of Isentropic Compression (*Kelvin*)
- **T_2** Ideal Temp at end of Isentropic Compression (*Kelvin*)
- **T_3** Ideal Temp at end of Isobaric Cooling (*Kelvin*)
- **T_4** Temperature at End of Isentropic Expansion (*Kelvin*)
- **T_f** Final Temperature (*Kelvin*)
- **T_i** Initial Temperature (*Kelvin*)
- **T_{ratio}** Temperature Ratio
- **u_{Fluid}** Fluid Velocity (*Meter per Second*)
- **v** Specific Volume (*Cubic Meter per Kilogram*)
- **V_f** Final Volume of System (*Cubic Meter*)













- V_i Initial Volume of System (Cubic Meter)
- V_{process} Velocity (Meter per Second)
- w Work Done (Kilojoule per Kilogram)
- W Work (Joule)
- W_b Isobaric Work (Joule)
- $W_{\text{per min}}$ Work Done per min (Kilojoule per Minute)
- γ Heat Capacity Ratio
- ΔS Change in Entropy (Joule per Kilogram K)
- ΔS_{CP} Entropy Change Constant Pressure (Joule per Kilogram K)
- ΔS_{CV} Entropy Change Constant Volume (Joule per Kilogram K)
- η Ram Efficiency



Constants, Functions, Measurements used

- **Constant:** **[R]**, 8.31446261815324 Joule / Kelvin * Mole
Universal gas constant
- **Function:** **ln**, ln(Number)
Natural logarithm function (base e)
- **Measurement:** **Weight** in Kilogram (kg)
Weight Unit Conversion 
- **Measurement:** **Time** in Minute (min)
Time Unit Conversion 
- **Measurement:** **Temperature** in Kelvin (K)
Temperature Unit Conversion 
- **Measurement:** **Amount of Substance** in Mole (mol)
Amount of Substance Unit Conversion 
- **Measurement:** **Volume** in Cubic Meter (m³)
Volume Unit Conversion 
- **Measurement:** **Area** in Square Meter (m²)
Area Unit Conversion 
- **Measurement:** **Pressure** in Bar (Bar), Pascal (Pa)
Pressure Unit Conversion 
- **Measurement:** **Speed** in Meter per Second (m/s)
Speed Unit Conversion 
- **Measurement:** **Energy** in Joule (J)
Energy Unit Conversion 
- **Measurement:** **Power** in Kilojoule per Minute (kJ/min), Btu (th) per Hour (Btu/h)
Power Unit Conversion 



- **Measurement: Heat of Combustion (per Mass)** in Kilojoule per Kilogram (kJ/kg)
Heat of Combustion (per Mass) Unit Conversion 
- **Measurement: Specific Heat Capacity** in Kilojoule per Kilogram per K (kJ/kg*K)
Specific Heat Capacity Unit Conversion 
- **Measurement: Mass Flow Rate** in Kilogram per Second (kg/s)
Mass Flow Rate Unit Conversion 
- **Measurement: Specific Volume** in Cubic Meter per Kilogram (m³/kg)
Specific Volume Unit Conversion 
- **Measurement: Specific Entropy** in Joule per Kilogram K (J/kg*K)
Specific Entropy Unit Conversion 
- **Measurement: Latent Heat** in Kilojoule per Kilogram (kJ/kg)
Latent Heat Unit Conversion 
- **Measurement: Rate of Heat Transfer** in Kilojoule per Minute (kJ/min)
Rate of Heat Transfer Unit Conversion 
- **Measurement: Specific Energy** in Kilojoule per Kilogram (kJ/kg)
Specific Energy Unit Conversion 
- **Measurement: Molar Specific Heat Capacity at Constant Pressure** in Joule Per Kelvin Per Mole (J/K* mol)
Molar Specific Heat Capacity at Constant Pressure Unit Conversion 
- **Measurement: Molar Specific Heat Capacity at Constant Volume** in Joule Per Kelvin Per Mole (J/K* mol)
Molar Specific Heat Capacity at Constant Volume Unit Conversion 



Check other formula lists

- [Refrigeration and Air Conditioning Formulas](#) 

Feel free to SHARE this document with your friends!

PDF Available in

[English](#) [Spanish](#) [French](#) [German](#) [Russian](#) [Italian](#) [Portuguese](#) [Polish](#) [Dutch](#)

9/22/2023 | 2:48:41 AM UTC

[Please leave your feedback here...](#)

