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# Vibrational Energy Levels Formulas

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# List of 15 Vibrational Energy Levels Formulas

## Vibrational Energy Levels

### 1) Anharmonicity Constant given Dissociation Energy

$$\text{fx } x_e = \frac{(\omega')^2}{4 \cdot D_e \cdot \omega'}$$

[Open Calculator !\[\]\(a870788d6ed9b8fd294b7654a8c8526b\_img.jpg\)](#)

$$\text{ex } 0.375 = \frac{(15/\text{m})^2}{4 \cdot 10\text{J} \cdot 15/\text{m}}$$

### 2) Dissociation Energy given Vibrational Wavenumber

$$\text{fx } D_e = \frac{\omega'^2}{4 \cdot x_e \cdot \omega'}$$

[Open Calculator !\[\]\(c50c8b7b2cc2cf9ff925edec0ee94c0d\_img.jpg\)](#)

$$\text{ex } 15.625\text{J} = \frac{(15/\text{m})^2}{4 \cdot 0.24 \cdot 15/\text{m}}$$

### 3) Dissociation Energy of Potential

$$\text{fx } D_{ae} = E_{vf} \cdot v_{\max}$$

[Open Calculator !\[\]\(f60b7a900783ac3fd531bfd9c111be6d\_img.jpg\)](#)

$$\text{ex } 550\text{J} = 100\text{J} \cdot 5.5$$



#### 4) Dissociation Energy of Potential using Zero Point Energy

$$fx \quad D_e = D_0 + E_0$$

[Open Calculator !\[\]\(cbe80b694ebd74fcfe136a095b608235\_img.jpg\)](#)

$$ex \quad 9J = 5J + 4J$$

#### 5) Energy of Vibrational Transitions

fx

[Open Calculator !\[\]\(3e2231b1ad3ca8da8658228c00dd08e0\_img.jpg\)](#)

$$E_t = \left( \left( v + \frac{1}{2} \right) - x_e \cdot \left( \left( v + \frac{1}{2} \right)^2 \right) \right) \cdot ([hP] \cdot v_{vib})$$

$$ex \quad 8.6E^{-34}J = \left( \left( 2 + \frac{1}{2} \right) - 0.24 \cdot \left( \left( 2 + \frac{1}{2} \right)^2 \right) \right) \cdot ([hP] \cdot 1.3Hz)$$

#### 6) Maximum Vibrational Quantum Number given Dissociation Energy

$$fx \quad v_m = \frac{D_e}{E_{vf}}$$

[Open Calculator !\[\]\(b792654f2cef9719eabeb6c5be00811e\_img.jpg\)](#)

$$ex \quad 0.1 = \frac{10J}{100J}$$

#### 7) Vibrational Energy

$$fx \quad E_t = \left( v + \frac{1}{2} \right) \cdot ([hP] \cdot v_{vib})$$

[Open Calculator !\[\]\(84f47badaad7772cd95667a7c387a639\_img.jpg\)](#)

$$ex \quad 2.2E^{-33}J = \left( 2 + \frac{1}{2} \right) \cdot ([hP] \cdot 1.3Hz)$$




8) Vibrational energy using Anharmonicity constant 

$$\text{fx } E_{xe} = \frac{(\omega')^2}{4 \cdot x_e \cdot \omega' \cdot v_{\max}}$$

Open Calculator 


$$\text{ex } 2.840909\text{J} = \frac{(15/\text{m})^2}{4 \cdot 0.24 \cdot 15/\text{m} \cdot 5.5}$$

9) Vibrational Energy using Dissociation Energy 

$$\text{fx } E_{DE} = \frac{D_e}{v_{\max}}$$

Open Calculator 


$$\text{ex } 1.818182\text{J} = \frac{10\text{J}}{5.5}$$

10) Vibrational Energy using Vibrational Wave Number 

$$\text{fx } E_{wn} = \left( v + \frac{1}{2} \right) \cdot \omega'$$

Open Calculator 

$$\text{ex } 37.5\text{J} = \left( 2 + \frac{1}{2} \right) \cdot 15/\text{m}$$


11) Vibrational Frequency given Vibrational Energy 

$$\text{fx } v_{ve} = \frac{E_{vf}}{v + \frac{1}{2}} \cdot [\text{hP}]$$

Open Calculator 

$$\text{ex } 2.7\text{E}^{-32}\text{Hz} = \frac{100\text{J}}{2 + \frac{1}{2}} \cdot [\text{hP}]$$



12) Vibrational Wavenumber given Vibrational Energy 

$$fx \quad \omega'_{ve} = \frac{E_{vf}}{v + \frac{1}{2}}$$

Open Calculator 

$$ex \quad 40 = \frac{100J}{2 + \frac{1}{2}}$$

13) Zero Point Dissociation Energy 

$$fx \quad D_0 = D_e - E_0$$

Open Calculator 

$$ex \quad 6J = 10J - 4J$$

14) Zero Point Energy 

$$fx \quad E_0 = \left( \frac{1}{2} \cdot \omega' \right) - \left( \frac{1}{4} \cdot x_e \cdot \omega' \right)$$

Open Calculator 

$$ex \quad 6.6J = \left( \frac{1}{2} \cdot 15/m \right) - \left( \frac{1}{4} \cdot 0.24 \cdot 15/m \right)$$

15) Zero Point Energy given Dissociation Energy 

$$fx \quad E_0 = D_e - D_0$$

Open Calculator 

$$ex \quad 5J = 10J - 5J$$






## Variables Used

- $D_0$  Zero Point Dissociation Energy (Joule)
- $D_{ae}$  Actual Dissociation Energy of Potential (Joule)
- $D_e$  Dissociation Energy of Potential (Joule)
- $E_0$  Zero Point Energy (Joule)
- $E_{DE}$  Vibrational Energy given DE (Joule)
- $E_t$  Vibrational Energy in Transition (Joule)
- $E_{vf}$  Vibrational Energy (Joule)
- $E_{wn}$  Vibrational Energy given wavenumber (Joule)
- $E_{xe}$  Vibrational Energy given xe constant (Joule)
- $v$  Vibrational Quantum Number
- $v_m$  Maximum Vibrational Number
- $v_{max}$  Max Vibrational Number
- $v_{ve}$  Vibrational Frequency given VE (Hertz)
- $v_{vib}$  Vibrational Frequency (Hertz)
- $x_e$  Anharmonicity Constant
- $\omega'$  Vibrational Wavenumber (1 per Meter)
- $\omega'_{ve}$  Vibrational Wavenumber given VE



## Constants, Functions, Measurements used

- **Constant:** [hP],  $6.626070040 \times 10^{-34}$  Kilogram Meter<sup>2</sup> / Second  
*Planck constant*
- **Measurement: Energy** in Joule (J)  
*Energy Unit Conversion* 
- **Measurement: Frequency** in Hertz (Hz)  
*Frequency Unit Conversion* 
- **Measurement: Wave Number** in 1 per Meter (1/m)  
*Wave Number Unit Conversion* 



## Check other formula lists

- **Vibrational Energy Levels Formulas** 

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