



[calculatoratoz.com](http://calculatoratoz.com)



[unitsconverters.com](http://unitsconverters.com)

## Important formulae on Clausius Model of Real Gas Formulas

Calculators!

Examples!

Conversions!

Bookmark [calculatoratoz.com](http://calculatoratoz.com), [unitsconverters.com](http://unitsconverters.com)

Widest Coverage of Calculators and Growing - **30,000+ Calculators!**  
Calculate With a Different Unit for Each Variable - **In built Unit Conversion!**  
Widest Collection of Measurements and Units - **250+ Measurements!**

Feel free to SHARE this document with your friends!

[Please leave your feedback here...](#)



## List of 19 Important formulae on Clausius Model of Real Gas Formulas

## Important formulae on Clausius Model of Real Gas ↗

## 1) Actual Pressure of Real Gas given Clausius Parameter a, Reduced and Critical Parameters ↗

$$\text{fx } P_a = \left( \frac{27 \cdot ([R]^2) \cdot (T_c^3)}{64 \cdot a} \right) \cdot P_r$$

Open Calculator ↗

$$\text{ex } 8.6E^8 Pa = \left( \frac{27 \cdot ([R]^2) \cdot ((154.4K)^3)}{64 \cdot 0.1} \right) \cdot 0.8$$

## 2) Actual Pressure of Real Gas given Clausius Parameter b, Reduced and Actual Parameters ↗

$$\text{fx } P_b = \left( \frac{[R] \cdot \left( \frac{T_{rg}}{T_r} \right)}{4 \cdot \left( \left( \frac{V_{real}}{V_r} \right) - b' \right)} \right) \cdot P_r$$

Open Calculator ↗

$$\text{ex } 21.56464 Pa = \left( \frac{[R] \cdot \left( \frac{300K}{10} \right)}{4 \cdot \left( \left( \frac{22L}{9.5L} \right) - 2.43E^{-3} \right)} \right) \cdot 0.8$$


## 3) Actual Pressure of Real Gas given Clausius Parameter c, Reduced and Actual Parameters ↗

$$\text{fx } P_c = \left( \frac{3 \cdot [R] \cdot \left( \frac{T_{rg}}{T_r} \right)}{8 \cdot \left( c + \left( \frac{V_{real}}{V_r} \right) \right)} \right) \cdot P_r$$

Open Calculator ↗

$$\text{ex } 32.31023 Pa = \left( \frac{3 \cdot [R] \cdot \left( \frac{300K}{10} \right)}{8 \cdot \left( 0.0002 + \left( \frac{22L}{9.5L} \right) \right)} \right) \cdot 0.8$$




4) Actual Temperature of Real Gas given Clausius Parameter a, Reduced and Actual Parameters 

[Open Calculator !\[\]\(4729e517bc6a7cd81c8025b9646574fb\_img.jpg\)](#)

$$\text{fx } T_{\text{RP}} = \left( \left( \frac{a \cdot 64 \cdot \left( \frac{P}{P_r} \right)}{27 \cdot ([R]^2)} \right)^{\frac{1}{3}} \right) \cdot T_r$$


$$\text{ex } 15.07935\text{K} = \left( \left( \frac{0.1 \cdot 64 \cdot \left( \frac{800\text{Pa}}{0.8} \right)}{27 \cdot ([R]^2)} \right)^{\frac{1}{3}} \right) \cdot 10$$

5) Actual Temperature of Real Gas using Critical and Reduced Temperature 

[Open Calculator !\[\]\(e474458956c9a37fbf9586ddb60a7fa1\_img.jpg\)](#)

$$\text{fx } T_{\text{RT}} = T_r \cdot T'_c$$


$$\text{ex } 1544\text{K} = 10 \cdot 154.4\text{K}$$

6) Actual Volume of Real Gas using Clausius Parameter b, Reduced and Critical Parameters 

[Open Calculator !\[\]\(4fe57c3593bf1b21d272ae7ac8dfaf77\_img.jpg\)](#)

$$\text{fx } V_{\text{real\_CP}} = \left( b' + \left( \frac{[R] \cdot T'_c}{4 \cdot P'_c} \right) \right) \cdot V_r$$

$$\text{ex } 0.023748\text{L} = \left( 2.43\text{E}^{-3} + \left( \frac{[R] \cdot 154.4\text{K}}{4 \cdot 4.6\text{E}^6\text{Pa}} \right) \right) \cdot 9.5\text{L}$$

7) Actual Volume of Real Gas using Clausius Parameter c, Reduced and Critical Parameters 

[Open Calculator !\[\]\(2bae76de5ebbd5c4d7d47162f1673734\_img.jpg\)](#)

$$\text{fx } V_{\text{real\_CP}} = \left( \left( \frac{3 \cdot [R] \cdot T'_c}{8 \cdot P'_c} \right) - c \right) \cdot V'_{m,r}$$

$$\text{ex } 2.137343\text{L} = \left( \left( \frac{3 \cdot [R] \cdot 647\text{K}}{8 \cdot 4.6\text{E}^6\text{Pa}} \right) - 0.0002 \right) \cdot 8.96$$


8) Clausius Parameter b given Reduced and Actual Parameters 

[Open Calculator !\[\]\(5d954b3e270654ad8ab0d5913161c03c\_img.jpg\)](#)

$$\text{fx } b_{\text{RP}} = \left( \frac{V_{\text{real}}}{V_r} \right) - \left( \frac{[R] \cdot \left( \frac{T_{\text{rg}}}{T_r} \right)}{4 \cdot \left( \frac{P}{P_r} \right)} \right)$$

$$\text{ex } 2.253431 = \left( \frac{22\text{L}}{9.5\text{L}} \right) - \left( \frac{[R] \cdot \left( \frac{300\text{K}}{10} \right)}{4 \cdot \left( \frac{800\text{Pa}}{0.8} \right)} \right)$$




9) Clausius Parameter c given Critical Parameters 

$$\text{fx } c_{CP} = \left( \frac{3 \cdot [R] \cdot T_c}{8 \cdot P_c} \right) - V_c$$

Open Calculator 


$$\text{ex } 9.243654 = \left( \frac{3 \cdot [R] \cdot 647K}{8 \cdot 218Pa} \right) - 10L$$

10) Critical Molar Volume of Real Gas using Clausius Equation given Reduced and Actual Parameters 

$$\text{fx } V_{RP} = \frac{\left( \frac{[R] \cdot T_{rg}}{P + \left( \frac{a}{T_{rg}} \right)} \right) + b'}{V'_{m,r}}$$

Open Calculator 

$$\text{ex } 0.348254m^3/mol = \frac{\left( \frac{[R] \cdot 300K}{800Pa + \left( \frac{0.1}{300K} \right)} \right) + 2.43E^{-3}}{8.96}$$

11) Critical Molar Volume using Clausius Equation given Actual and Critical Parameters 

$$\text{fx } V_{RP} = \frac{\left( \frac{[R] \cdot T_{rg}}{P + \left( \frac{a}{T_{rg}} \right)} \right) + b'}{V_m}$$

Open Calculator 


$$\text{ex } 0.139301m^3/mol = \frac{\left( \frac{[R] \cdot 300K}{800Pa + \left( \frac{0.1}{300K} \right)} \right) + 2.43E^{-3}}{22.4m^3/mol}$$

12) Critical Pressure of Real Gas using Actual and Reduced Pressure 

$$\text{fx } P_{CP} = \frac{P}{P_r}$$

Open Calculator 

$$\text{ex } 1000Pa = \frac{800Pa}{0.8}$$

13) Critical Temperature given Clausius Parameter c, Reduced and Actual Parameters 

$$\text{fx } T_{c\_RP} = \frac{\left( c + \left( \frac{V_{real}}{V_r} \right) \right) \cdot 8 \cdot \left( \frac{P}{P_r} \right)}{3 \cdot [R]}$$

Open Calculator 

$$\text{ex } 742.7987K = \frac{\left( 0.0002 + \left( \frac{22L}{9.5L} \right) \right) \cdot 8 \cdot \left( \frac{800Pa}{0.8} \right)}{3 \cdot [R]}$$



14) Molar Volume of Real Gas using Clausius Equation 

[Open Calculator !\[\]\(bd1a142de767a21e5362c595f844a4ff\_img.jpg\)](#)

$$\text{fx } V_{m\_CE} = \left( \frac{[R] \cdot T_{rg}}{p + \left( \frac{a}{T_{rg}} \right)} \right) + b'$$


$$\text{ex } 3.120352\text{m}^3/\text{mol} = \left( \frac{[R] \cdot 300\text{K}}{800\text{Pa} + \left( \frac{0.1}{300\text{K}} \right)} \right) + 2.43\text{E}^{-3}$$

15) Reduced Pressure of Real Gas using Actual and Critical Pressure 

[Open Calculator !\[\]\(830769b31eeeaca920791081939ff8ba\_img.jpg\)](#)

$$\text{fx } P_{r\_AP\_RP} = \frac{P_{rg}}{P'_c}$$

$$\text{ex } 0.002203 = \frac{10132\text{Pa}}{4.6\text{E}^6\text{Pa}}$$

16) Reduced Temperature of Real Gas using Clausius Equation given Reduced and Actual Parameters 

[Open Calculator !\[\]\(47734e4656765d20df4fdbd5b7aff048\_img.jpg\)](#)

$$\text{fx } T_{r\_RP\_AP} = \frac{\left( p + \left( \frac{a}{(V_m+c)^2} \right) \right) \cdot \left( \frac{V_m-b'}{[R]} \right)}{T_{rg}}$$

$$\text{ex } 7.183491 = \frac{\left( 800\text{Pa} + \left( \frac{0.1}{(22.4\text{m}^3/\text{mol}+0.0002)^2} \right) \right) \cdot \left( \frac{22.4\text{m}^3/\text{mol}-2.43\text{E}^{-3}}{[R]} \right)}{300\text{K}}$$


17) Reduced Volume of Real Gas given Clausius Parameter c, Reduced and Actual Parameters 

[Open Calculator !\[\]\(41aea2746216b27a6939d696d8e035da\_img.jpg\)](#)

$$\text{fx } V_{r\_RP\_AP} = \frac{V_{real}}{\left( \frac{3 \cdot [R] \cdot \left( \frac{T_{real}}{T_r} \right)}{8 \cdot \left( \frac{P_{real}}{P_r} \right)} \right) - c}$$


$$\text{ex } 0.029702 = \frac{22\text{L}}{\left( \frac{3 \cdot [R] \cdot \left( \frac{300\text{K}}{10} \right)}{8 \cdot \left( \frac{101\text{Pa}}{0.5} \right)} \right) - 0.0002}$$



18) Temperature of Real Gas using Clausius Equation [Open Calculator](#) 

$$\text{fx } T_{\text{CE}} = \left( p + \left( \frac{a}{(V_m + c)^2} \right) \right) \cdot \left( \frac{V_m - b'}{[R]} \right)$$

$$\text{ex } 2155.047\text{K} = \left( 800\text{Pa} + \left( \frac{0.1}{((22.4\text{m}^3/\text{mol} + 0.0002)^2)} \right) \right) \cdot \left( \frac{22.4\text{m}^3/\text{mol} - 2.43\text{E}^{-3}}{[R]} \right)$$

19) Temperature of Real Gas using Clausius Equation given Reduced and Critical Parameters [Open Calculator](#) 

$$\text{fx } T_{\text{CE}} = \left( (P_r \cdot P'_c) + \left( \frac{a}{(((V'_{m,r} \cdot V_{m,c}) + c)^2)} \right) \right) \cdot \left( \frac{(V'_{m,r} \cdot V_{m,c}) - b'}{[R]} \right)$$

$$\text{ex } 4.6\text{E}^7\text{K} = \left( (0.8 \cdot 4.6\text{E}^6\text{Pa}) + \left( \frac{0.1}{(((8.96 \cdot 11.5\text{m}^3/\text{mol}) + 0.0002)^2)} \right) \right) \cdot \left( \frac{(8.96 \cdot 11.5\text{m}^3/\text{mol}) - 2.43\text{E}^{\wedge}}{[R]} \right)$$



## Variables Used

- **a** Clausius Parameter a
- **b'** Clausius Parameter b for Real Gas
- **b<sub>RP</sub>** Clausius Parameter b given RP
- **c** Clausius Parameter c
- **c<sub>CP</sub>** Clausius Parameter c given CP
- **p** Pressure (Pascal)
- **P<sub>c</sub>** Critical Pressure (Pascal)
- **P'<sub>c</sub>** Critical Pressure of Real Gas (Pascal)
- **P<sub>CP</sub>** Critical Pressure given RP (Pascal)
- **P<sub>r</sub>** Reduced Pressure
- **P<sub>r\_AP\_RP</sub>** Reduced Pressure given RP AP
- **P<sub>real</sub>** Real Gas Pressure (Pascal)
- **P<sub>rg</sub>** Pressure of Gas (Pascal)
- **P<sub>a</sub>** Pressure given a (Pascal)
- **P<sub>b</sub>** Pressure given b (Pascal)
- **P<sub>c</sub>** Pressure given c (Pascal)
- **T<sub>c</sub>** Critical Temperature (Kelvin)
- **T'<sub>c</sub>** Critical Temperature For Clausius Model (Kelvin)
- **T<sub>c\_RP</sub>** Critical temperature given RP (Kelvin)
- **T<sub>CE</sub>** Temperature given CE (Kelvin)
- **T<sub>r</sub>** Reduced Temperature
- **T<sub>r\_RP\_AP</sub>** Reduced Temperature given RP AP
- **T<sub>real</sub>** Real Gas Temperature (Kelvin)
- **T<sub>rg</sub>** Temperature of Real Gas (Kelvin)
- **T<sub>RP</sub>** Temperature given RP (Kelvin)
- **T<sub>RT</sub>** Temperature given RT (Kelvin)
- **V<sub>c</sub>** Critical Volume (Liter)
- **V<sub>m</sub>** Molar Volume (Cubic Meter per Mole)
- **V<sub>m,c</sub>** Critical Molar Volume (Cubic Meter per Mole)
- **V'<sub>m,r</sub>** Reduced Molar Volume for Real Gas
- **V<sub>m\_CE</sub>** Molar Volume given CE (Cubic Meter per Mole)
- **V<sub>r</sub>** Reduced Volume (Liter)
- **V<sub>r\_RP\_AP</sub>** Reduced Volume given RP AP







- $V_{\text{real}}$  Volume of Real Gas (Liter)
- $V_{\text{real\_CP}}$  Volume of Real Gas given CP (Liter)
- $V_{\text{RP}}$  Critical Molar Volume given RP (Cubic Meter per Mole)










## Constants, Functions, Measurements used

- **Constant:** **[R]**, 8.31446261815324 Joule / Kelvin \* Mole  
*Universal gas constant*
- **Measurement:** **Temperature** in Kelvin (K)  
*Temperature Unit Conversion* 
- **Measurement:** **Volume** in Liter (L)  
*Volume Unit Conversion* 
- **Measurement:** **Pressure** in Pascal (Pa)  
*Pressure Unit Conversion* 
- **Measurement:** **Molar Magnetic Susceptibility** in Cubic Meter per Mole (m<sup>3</sup>/mol)  
*Molar Magnetic Susceptibility Unit Conversion* 



## Check other formula lists

- [Actual Pressure of Real Gas Formulas](#) 
- [Actual Temperature of Real Gas Formulas](#) 
- [Actual Volume of Real Gas Formulas](#) 
- [Clausius Parameter Formulas](#) 
- [Critical Pressure Formulas](#) 
- [Critical Temperature Formulas](#) 
- [Important formulae on Clausius Model of Real Gas Formulas](#) 
- [Reduced Pressure of Real Gas Formulas](#) 
- [Reduced Temperature of Real Gas Formulas](#) 
- [Reduced Volume Formulas](#) 

Feel free to SHARE this document with your friends!

## PDF Available in

[English](#) [Spanish](#) [French](#) [German](#) [Russian](#) [Italian](#) [Portuguese](#) [Polish](#) [Dutch](#)

1/1/2024 | 5:01:53 AM UTC

[Please leave your feedback here...](#)

