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Important Formulas on Enzyme Kinetics

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List of 26 Important Formulas on Enzyme Kinetics

Important Formulas on Enzyme Kinetics

1) Catalytic Rate Constant from Michaelis Menten Kinetics Equation

$$fx \quad k_{cat_MM} = \frac{V_0 \cdot (K_M + S)}{([E_0]) \cdot S}$$

[Open Calculator !\[\]\(a870788d6ed9b8fd294b7654a8c8526b_img.jpg\)](#)

$$ex \quad 0.0135s^{-1} = \frac{0.45mol/L*s \cdot (3mol/L + 1.5mol/L)}{100mol/L \cdot 1.5mol/L}$$

2) Catalytic Rate Constant if Substrate Concentration is higher than Michaelis Constant

$$fx \quad k_{cat} = \frac{V_{max}}{[E_0]}$$

[Open Calculator !\[\]\(c50c8b7b2cc2cf9ff925edec0ee94c0d_img.jpg\)](#)

$$ex \quad 0.4s^{-1} = \frac{40mol/L*s}{100mol/L}$$


3) Dissociation Constant of Enzyme given Modifying Factor of Enzyme

$$fx \quad K_{ei} = \frac{I}{\alpha - 1}$$

[Open Calculator !\[\]\(f60b7a900783ac3fd531bfd9c111be6d_img.jpg\)](#)

$$ex \quad 2.25mol/L = \frac{9mol/L}{5 - 1}$$



4) Dissociation Rate Constant in Enzymatic Reaction Mechanism 

$$fx \quad K_D = \frac{k_r}{k_f}$$

[Open Calculator !\[\]\(cbe80b694ebd74fcfe136a095b608235_img.jpg\)](#)

$$ex \quad 2.898551 \text{mol/L} = \frac{20 \text{mol/L} \cdot s}{6.9 \text{s}^{-1}}$$

5) Enzyme Catalyst Concentration given Forward, Reverse, and Catalytic Rate Constants 

$$fx \quad E = \frac{(k_r + k_{cat}) \cdot ES}{k_f \cdot S}$$

[Open Calculator !\[\]\(3e2231b1ad3ca8da8658228c00dd08e0_img.jpg\)](#)

$$ex \quad 19.3243 \text{mol/L} = \frac{(20 \text{mol/L} \cdot s + 0.65 \text{s}^{-1}) \cdot 10 \text{mol/L}}{6.9 \text{s}^{-1} \cdot 1.5 \text{mol/L}}$$

6) Enzyme Concentration from Michaelis Menten Kinetics equation 

$$fx \quad ([E_i]) = \frac{V_0 \cdot (K_M + S)}{k_{cat} \cdot S}$$

[Open Calculator !\[\]\(0d5ec72f61334709c3fc9450209b754f_img.jpg\)](#)

$$ex \quad 2.076923 \text{mol/L} = \frac{0.45 \text{mol/L} \cdot s \cdot (3 \text{mol/L} + 1.5 \text{mol/L})}{0.65 \text{s}^{-1} \cdot 1.5 \text{mol/L}}$$



7) Enzyme Substrate Complex Concentration for Competitive Inhibition of Enzyme Catalysis

$$\text{fx } ES = \frac{S \cdot ([E_0])}{K_M \cdot \left(1 + \left(\frac{I}{K_i}\right)\right) + S}$$

[Open Calculator !\[\]\(e78f798d4ea5c530c9db49e7d26e6b95_img.jpg\)](#)

$$\text{ex } 25.333333\text{mol/L} = \frac{1.5\text{mol/L} \cdot 100\text{mol/L}}{3\text{mol/L} \cdot \left(1 + \left(\frac{9\text{mol/L}}{19\text{mol/L}}\right)\right) + 1.5\text{mol/L}}$$

8) Final Rate Constant for Competitive Inhibition of Enzyme Catalysis

$$\text{fx } k_{\text{final}} = \frac{V_0 \cdot \left(K_M \cdot \left(1 + \left(\frac{I}{K_i}\right)\right) + S\right)}{([E_0]) \cdot S}$$

[Open Calculator !\[\]\(05be7c7a8995decd503647c99211f7c2_img.jpg\)](#)

$$\text{ex } 0.017763\text{s}^{-1} = \frac{0.45\text{mol/L} \cdot \text{s} \cdot \left(3\text{mol/L} \cdot \left(1 + \left(\frac{9\text{mol/L}}{19\text{mol/L}}\right)\right) + 1.5\text{mol/L}\right)}{100\text{mol/L} \cdot 1.5\text{mol/L}}$$

9) Forward Rate Constant given Dissociation Rate Constant

$$\text{fx } k_f = \left(\frac{k_r}{K_D}\right)$$

[Open Calculator !\[\]\(fe3aebe81acea8d45108cd2768939da7_img.jpg\)](#)

$$\text{ex } 3.508772\text{s}^{-1} = \left(\frac{20\text{mol/L} \cdot \text{s}}{5.7\text{mol/L}}\right)$$



10) Inhibitor Concentration for Competitive Inhibition of Enzyme Catalysis 

$$\text{fx } I_{\text{IEC}} = \left(\left(\frac{\left(\frac{k_2 \cdot ([E_0]) \cdot S}{V_0} \right) - S}{K_M} \right) - 1 \right) \cdot K_i$$

Open Calculator 

ex


$$48527.06 \text{ mol/L} = \left(\left(\frac{\left(\frac{23 \text{ s}^{-1} \cdot 100 \text{ mol/L} \cdot 1.5 \text{ mol/L}}{0.45 \text{ mol/L} \cdot \text{s}} \right) - 1.5 \text{ mol/L}}{3 \text{ mol/L}} \right) - 1 \right) \cdot 19 \text{ mol/L}$$

11) Inhibitor Concentration given Apparent Initial Enzyme Concentration 

$$\text{fx } I_{\text{CI}} = \left(\left(\frac{[E_0]}{E_0^{\text{app}}} \right) - 1 \right) \cdot K_i$$

Open Calculator 

$$\text{ex } 31647.67 \text{ mol/L} = \left(\left(\frac{100 \text{ mol/L}}{0.06 \text{ mol/L}} \right) - 1 \right) \cdot 19 \text{ mol/L}$$

12) Inhibitor Concentration given Enzyme Substrate Modifying Factor 

$$\text{fx } I = (\alpha' - 1) \cdot (K_i')$$

Open Calculator 

$$\text{ex } 15 \text{ mol/L} = (2 - 1) \cdot 15 \text{ mol/L}$$



13) Inhibitor Concentration in Competitive Inhibition given Maximum Rate of System

$$\text{fx } I_{\max} = \left(\left(\left(\frac{V_{\max} \cdot S}{V_0} \right) - S \right) \right) \cdot K_M - 1 \cdot K_i$$

[Open Calculator !\[\]\(d3fb9f94af8b26d1c844efa9a98805b0_img.jpg\)](#)

$$\text{ex } 815.9444\text{mol/L} = \left(\left(\left(\frac{40\text{mol/L} \cdot \text{s} \cdot 1.5\text{mol/L}}{0.45\text{mol/L} \cdot \text{s}} \right) - 1.5\text{mol/L} \right) \right) \cdot 3\text{mol/L} - 1 \cdot 19\text{mol/L}$$

14) Initial Concentration of Enzyme in presence of Inhibitor by Enzyme Conservation Law

$$\text{fx } ([E_{\text{initial}}]) = (E + ES + EI)$$

[Open Calculator !\[\]\(e1d6102fe77919492c04879c8450f1f5_img.jpg\)](#)

$$\text{ex } 64\text{mol/L} = (25\text{mol/L} + 10\text{mol/L} + 29\text{mol/L})$$


15) Initial Enzyme Concentration if Substrate Concentration is Higher than Michaelis Constant

$$\text{fx } ([E_{\text{initial}}]) = \frac{V_{\max}}{k_{\text{cat}}}$$

[Open Calculator !\[\]\(ab4e2b3fc7e7887b7a72f548aa6f5e60_img.jpg\)](#)

$$\text{ex } 61.53846\text{mol/L} = \frac{40\text{mol/L} \cdot \text{s}}{0.65\text{s}^{-1}}$$



16) Initial Enzyme Concentration given Dissociation Rate Constant 

$$fx \quad ([E]_{\text{initial}}) = \frac{ES \cdot (K_D + S)}{S}$$

Open Calculator 


$$ex \quad 48\text{mol/L} = \frac{10\text{mol/L} \cdot (5.7\text{mol/L} + 1.5\text{mol/L})}{1.5\text{mol/L}}$$

17) Initial Rate in Competitive Inhibition given Maximum Rate of system 

$$fx \quad V_{CI} = \frac{V_{\text{max}} \cdot S}{K_M \cdot \left(1 + \left(\frac{I}{K_i}\right)\right) + S}$$

Open Calculator 

$$ex \quad 10.13333\text{mol/L*s} = \frac{40\text{mol/L*s} \cdot 1.5\text{mol/L}}{3\text{mol/L} \cdot \left(1 + \left(\frac{9\text{mol/L}}{19\text{mol/L}}\right)\right) + 1.5\text{mol/L}}$$

18) Initial Rate of System given Rate Constant and Enzyme Substrate Complex Concentration 

$$fx \quad V_{RC} = k_2 \cdot ES$$

Open Calculator 

$$ex \quad 230\text{mol/L*s} = 23\text{s}^{-1} \cdot 10\text{mol/L}$$


19) Initial Reaction Rate given Dissociation Rate Constant 

$$fx \quad V_{DRC} = \frac{V_{\text{max}} \cdot S}{K_D + S}$$

Open Calculator 

$$ex \quad 8.333333\text{mol/L*s} = \frac{40\text{mol/L*s} \cdot 1.5\text{mol/L}}{5.7\text{mol/L} + 1.5\text{mol/L}}$$



20) Maximum Rate given Dissociation Rate Constant 

$$\text{fx } V_{\text{max_DRC}} = \frac{V_0 \cdot (K_D + S)}{S}$$

Open Calculator 

$$\text{ex } 2.16\text{mol/L*s} = \frac{0.45\text{mol/L*s} \cdot (5.7\text{mol/L} + 1.5\text{mol/L})}{1.5\text{mol/L}}$$

21) Maximum Rate if Substrate Concentration is Higher than Michaelis Constant 

$$\text{fx } V_{\text{max}} = k_{\text{cat}} \cdot ([E_0])$$

Open Calculator 


$$\text{ex } 65\text{mol/L*s} = 0.65\text{s}^{-1} \cdot 100\text{mol/L}$$

22) Maximum Rate in presence of Noncompetitive Inhibitor 

$$\text{fx } V_{\text{max}} = \left(V_{\text{max}}^{\text{app}} \cdot \left(1 + \left(\frac{I}{K_i} \right) \right) \right)$$

Open Calculator 

$$\text{ex } 30.94737\text{mol/L*s} = \left(21\text{mol/L*s} \cdot \left(1 + \left(\frac{9\text{mol/L}}{19\text{mol/L}} \right) \right) \right)$$

23) Michaelis Constant given Forward, Reverse, and Catalytic Rate Constants 

$$\text{fx } K_M = \frac{k_r + k_{\text{cat}}}{k_f}$$

Open Calculator 

$$\text{ex } 2.898645\text{mol/L} = \frac{20\text{mol/L*s} + 0.65\text{s}^{-1}}{6.9\text{s}^{-1}}$$



24) Michaelis Constant in Competitive Inhibition given Enzyme Substrate Complex Concentration

[Open Calculator !\[\]\(666e09182d4cd268646ea700ea60dcdf_img.jpg\)](#)

$$\text{fx } K_M = \frac{\left(\frac{([E_0]) \cdot S}{ES} \right) - S}{1 + \left(\frac{I}{K_i} \right)}$$

$$\text{ex } 9.160714 \text{mol/L} = \frac{\left(\frac{100 \text{mol/L} \cdot 1.5 \text{mol/L}}{10 \text{mol/L}} \right) - 1.5 \text{mol/L}}{1 + \left(\frac{9 \text{mol/L}}{19 \text{mol/L}} \right)}$$

25) Modifying Factor of Enzyme Substrate Complex

[Open Calculator !\[\]\(003082e50e3009141f59bd5df831749f_img.jpg\)](#)

$$\text{fx } \alpha' = 1 + \left(\frac{I}{K_i'} \right)$$

$$\text{ex } 1.6 = 1 + \left(\frac{9 \text{mol/L}}{15 \text{mol/L}} \right)$$

26) Substrate Concentration given Catalytic Rate Constant and Initial Enzyme Concentration

[Open Calculator !\[\]\(d3102649f02e825ddb76dc3de0190154_img.jpg\)](#)

$$\text{fx } S_o = \frac{K_M \cdot V_0}{(k_{\text{cat}} \cdot ([E_0])) - V_0}$$

$$\text{ex } 0.020914 \text{mol/L} = \frac{3 \text{mol/L} \cdot 0.45 \text{mol/L} \cdot \text{s}}{(0.65 \text{s}^{-1} \cdot 100 \text{mol/L}) - 0.45 \text{mol/L} \cdot \text{s}}$$



Variables Used




- $[E_0]$ Initial Enzyme Concentration (Mole per Liter)
- $[E_i]$ Initial Concentration of Enzyme (Mole per Liter)
- $[E_{\text{initial}}]$ Enzyme Concentration Initially (Mole per Liter)
- E Catalyst Concentration (Mole per Liter)
- $E0^{\text{app}}$ Apparent Initial Enzyme Concentration (Mole per Liter)
- EI Enzyme Inhibitor Complex Concentration (Mole per Liter)
- ES Enzyme Substrate Complex Concentration (Mole per Liter)
- I Inhibitor Concentration (Mole per Liter)
- I_{CI} Inhibitor Concentration for CI (Mole per Liter)
- I_{IEC} Inhibitor Concentration given IEC (Mole per Liter)
- I_{max} Inhibitor Concentration given Max Rate (Mole per Liter)
- k_2 Final Rate Constant (1 Per Second)
- k_{cat} Catalytic Rate Constant (1 Per Second)
- $k_{\text{cat_MM}}$ Catalytic Rate Constant for MM (1 Per Second)
- K_D Dissociation Rate Constant (Mole per Liter)
- K_{ei} Enzyme Inhibitor Dissociation Constant given MF (Mole per Liter)
- k_f Forward Rate Constant (1 Per Second)
- k_{final} Final Rate Constant for Catalysis (1 Per Second)
- K_i Enzyme Inhibitor Dissociation Constant (Mole per Liter)
- K_i' Enzyme Substrate Dissociation Constant (Mole per Liter)
- K_M Michaelis Constant (Mole per Liter)
- k_r Reverse Rate Constant (Mole per Liter Second)
- S Substrate Concentration (Mole per Liter)



- S_0 Concentration of Substrate (Mole per Liter)
- V_0 Initial Reaction Rate (Mole per Liter Second)
- V_{CI} Initial Reaction Rate in CI (Mole per Liter Second)
- V_{DRC} Initial Reaction Rate given DRC (Mole per Liter Second)
- V_{max} Maximum Rate (Mole per Liter Second)
- V_{max_DRC} Maximum Rate given DRC (Mole per Liter Second)
- V_{RC} Initial Reaction Rate given RC (Mole per Liter Second)
- V_{max}^{app} Apparent Maximum Rate (Mole per Liter Second)
- α Enzyme Modifying Factor
- α' Enzyme Substrate Modifying Factor













Constants, Functions, Measurements used

- **Measurement: Molar Concentration** in Mole per Liter (mol/L)
Molar Concentration Unit Conversion 
- **Measurement: Reaction Rate** in Mole per Liter Second (mol/L*s)
Reaction Rate Unit Conversion 
- **Measurement: First Order Reaction Rate Constant** in 1 Per Second (s^{-1})
First Order Reaction Rate Constant Unit Conversion 



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